Redefining Benchmarks for the Electrocatalytic Reduction of CO₂ to Formate

Scientific Achievement

We have discovered an Ir(III) pincer dihydride complex that electrocatalytically reduces CO₂ to formate with a high efficiency (85%) at a rapid rate (20/s) with little to no competitive CO production (<1%).

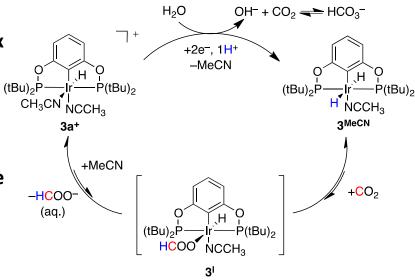
Significance and Impact

Until now, electrocatalytic reduction of CO₂ to formate using molecular catalysts was largely an inefficient process or was accompanied by the nonselective formation of CO and H₂. The Ir pincer catalyst described in this work overcomes many of these hurdles.

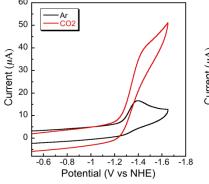
Research Details

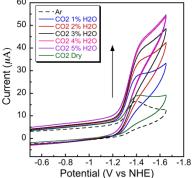
- Five-coordinate, 16-electron Ir(III) pincer dihydrides readily insert CO₂ and are electrocatalysts for selective reduction of CO₂ to formate.
- Electrocatalytic CO₂ reduction in acetonitrile occurs at ca.
 -1.25 V vs NHE in the presence of water (5%).
- Water plays a critical role in decreasing the CO₂ reduction potential and increasing the catalytic rate.

Peng Kang, Chen Cheng, Zuofeng Chen, Cynthia K. Schauer, Thomas J. Meyer, Maurice Brookhart. *J. Am. Chem. Soc.* **2012**, *134*, *5500*..



Overall: $2CO_2 + H_2O + 2e^- \rightarrow HCOO^- + HCO_3^-$





Top: Proposed mechanism for electrocatalytic CO₂ reduction in 5% H₂O/acetonitrile. **Bottom**: Effect of increasing [H₂O].

Work was performed at the University of North Carolina at Chapel Hill









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